Russian Journal of Organic Chemistry, Vol. 37, No. 11, 2001, pp. 1621–1628. Translated from Zhurnal Organicheskoi Khimii, Vol. 37, No. 11, 2001, pp. 1693–1699. Original Russian Text Copyright © 2001 by Furin, Ki-Wan Chi, Protsuk, Lopyrev.

## Reactions of Perfluoro(5-aza-4-nonene) with Hydrazine and Some Its Derivatives

G. G. Furin<sup>1</sup>, Ki-Wan Chi<sup>2</sup>, N. I. Protsuk<sup>3</sup>, and V. A. Lopyrev<sup>3</sup>

<sup>1</sup> Vorozhtsov Novosibirsk Institute of Organic Chemistry, Siberian Division, Russian Academy of Sciences, pr. Akademika Lavrent'eva 9, Novosibirsk, 630090 Russia e-mail: benzol@nioch.nsc.ru

<sup>2</sup> Ulsan University, South Korea

<sup>3</sup> Favorskii Irkutsk Institute of Chemistry, Siberian Division, Russian Academy of Sciences, ul. Favorskogo 1, Irkutsk, 664033 Russia

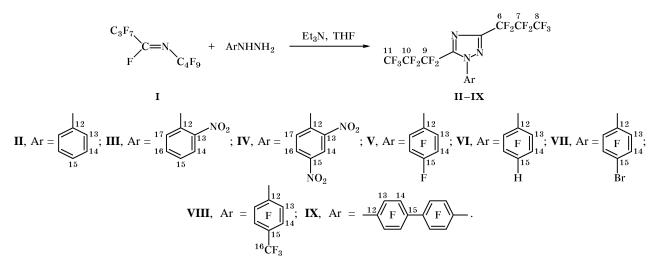
Received October 6, 2000

**Abstract**—The reaction of perfluoro(5-aza-4-nonene) with hydrazine hydrate in tetrahydrofuran at  $0-20^{\circ}$ C in the presence of triethylamine yields 2,5-bis(heptafluoropropyl)-1*H*-1,2,4-triazole; under similar conditions perfluoro(5-aza-4-nonene) reacts with arylhydrazines to form 1-aryl-3,5-bis(heptafluoropropyl)-1,2,4-triazoles.

1-Aryl-substituted azoles having perfluoroalkyl groups attract interest as intermediate products in the synthesis of potential biologically active compounds which can be used in medicine and agriculture [1, 2]. As a rule, such products are obtained via introduction of perfluoroalkyl groups into heteroring by various methods [3, 4]. In the recent years, a new approach was developed on the basis of reactions of commercially available perfluoroolefins with difunctional nucleophilic reagents [5–12]. We previously showed [13] that perfluoro(2-methyl-2-pentene) reacts with arylhydrazines in the presence of triethylamine to

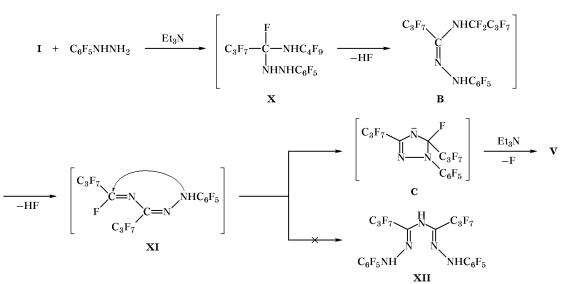
afford 1-aryl-substituted pyrazole derivatives. The key point of the process is formation of intermediates containing a C=C-C=N-N conjugated bond sequence. Intramolecular cyclization of such intermediates leads to formation of five-membered heterorings. While further developing the above approach, the present work was aimed at finding an initial fluorinated olefin whose reaction with arylhydrazines would give an intermediate having a C=N-N=N-Nconjugated bond system. Our previous studies [14] revealed that perfluoro(5-aza-4-nonene) (I) reacts with aliphatic and aromatic amidines to form *sym*-triazine

Scheme 1.



1070-4280/01/3711-1621 \$25.00 ©2001 MAIK "Nauka/Interperiodica"





derivatives. We now focused on the reactions of **I** with hydrazine hydrate and some its derivatives.

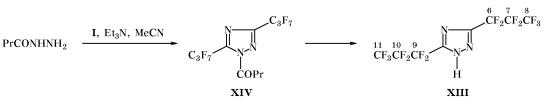
Perfluoro(5-aza-4-nonene) (I) reacted with arylhydrazines in tetrahydrofuran in the presence of triethylamine under very mild conditions, yielding 1-aryl-3,5-bis(heptafluoropropyl)-1,2,4-triazoles II-IX (Scheme 1). Presumably, the final products are formed as follows. Nucleophilic attack on compound I by the NH<sub>2</sub> group of pentafluorophenylhydrazine gives intermediate **X** which is capable of losing hydrogen fluoride by the action of triethylamine. Compound XI could give rise to two processes. The first of these includes replacement of fluorine at the C=N bond by hydrazino group and subsequent formation of bishydrazone **XII**. The second pathway is intramolecular nucleophilic cyclization with formation of 1,2,4-triazole derivative V (Scheme 2). The nature of the aryl group weakly affects the process. In the reaction with 4,4'-dihydrazinooctafluorobiphenyl two triazole rings were formed from the two hydrazine moieties (compound IX).

Propionyl hydrazine reacted with flouoroolefin **I** in the presence of triethylamine in acetonitrile to give bis(heptafluoropropyl)-1H-1,2,4-triazole (**XIII**) (Scheme 3), presumably as a result of elimination of the *N*-propionyl group from intermediate **XIV**. No dihydrazide formation was observed.

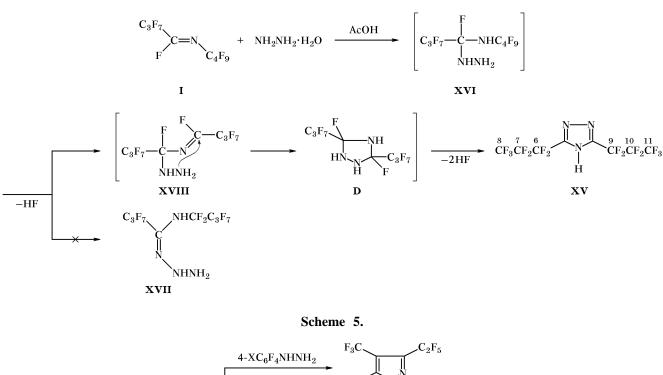
By reaction of compound **I** with hydrazine hydrate in acetic acid we obtained cyclization product **XV** in 42% yield. A possible mechanism of this reaction is shown in Scheme 4. Elimination of hydrogen fluoride from the hydrazino group of primary adduct **XVI** does not occur due to its high basicity, and compound **XVII** is not formed. Instead, HF molecule is released from the  $-NH-CF_2C_3F_7$  moiety to give intermediate **XVIII**. Intramolecular cyclization of the latter, followed by loss of two hydrogen fluoride molecules (from intermediate **D**), leads to formation of final product **XV**.

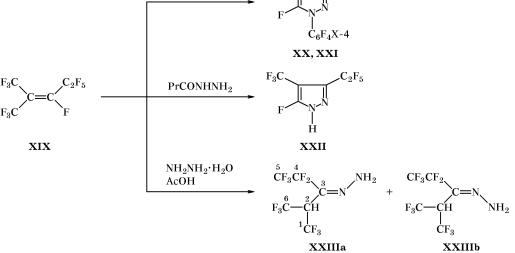
Perfluoro(2-methyl-2-pentene) (XIX) behaves like compound I in reactions with 2,3,5,6-tetrafluorophenylhydrazine, 4-bromo-2,3,5,6-tetrafluorophenylhydrazine, and propionylhydrazine. The corresponding pyrazole derivatives XX–XXII were thus obtained (Scheme 5). On the other hand, the reaction of XIX with hydrazine hydrate under the same conditions gave isomeric hydrazones XXIIIa and XXIIIb. Our attempts to effect cyclization of compounds XXIIIa and XXIIIb by the action of triethylamine in acetonitrile resulted in formation of isomeric perfluoro-











XX, X = H; XXI, X = Br.

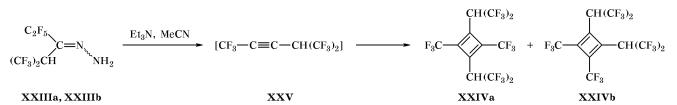
alkylcyclobutadienes **XXIVa** and **XXIVb** at a ratio of 2.3:1 (yield 63%) instead of the expected pyrazole derivatives (Scheme 6). Products **XXIVa** and **XXIVb** are likely to be formed via dimerization of intermediate alkyne **XXV**.

## **EXPERIMENTAL**

The <sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR spectra were recorded on a Bruker WP-400SY spectrometer at 400, 100, and 188 MHz, respectively; hexamethyldisiloxane and hexafluorobenzene were used as internal references  $({}^{13}C{-}^{1}H$  coupling constants were not measured). The IR spectra were obtained on a Specord M-80 instrument in CCl<sub>4</sub>. The mass spectra (70 eV) were run on a VG 707 OE GC–MS system. Perfluoro(5-aza-4-nonene) (I) was synthesized by the procedure described in [15].

**Typical procedure for synthesis of fluorinated** *N*-aryl-1,2,4-triazoles. A solution of 1 equiv of compound I in 10 ml of THF was added over a period of 15 min to a solution of 1 equiv of arylhydrazine and

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 37 No. 11 2001



3 equiv of triethylamine in 40 ml of THF on stirring at 0°C. The mixture was stirred for 1 h at 0°C, for 2 h at room temperature, and for 2 h at 50°C. It was then poured into 100 ml of water and extracted with methylene chloride ( $3 \times 70$  ml). The extract was dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated, and the residue was distilled under reduced pressure or was subjected to column chromatography using hexane–methylene chloride as eluent.

3,5-Bis(heptafluoropropyl)-1-phenyl-1,2,4-triazole (II) was synthesized from 2.48 g (23 mmol) of phenylhydrazine and 9.96 g (23 mmol) of perfluoroolefin I. Yield 9.2 g (83%); yellow liquid, bp 81-82°C (0.3 mm). IR spectrum, v, cm<sup>-1</sup>: 1658, 1520 (C=N-C=C), 1338 (C-N), 1236–1126 (C-F). <sup>1</sup>H NMR spectrum, δ, ppm: 7.26 (13-H), 7.36 (14-H), 7.31 (15-H). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 153.0 (C<sup>3</sup>,  ${}^{2}J_{C,F} = 29.5$  Hz), 144.6 (C<sup>5</sup>,  ${}^{2}J_{C,F} = 30.1$  Hz), 135.5 (C<sup>12</sup>), 130.1 (C<sup>15</sup>), 128.8 (C<sup>14</sup>), 125.0 (C<sup>13</sup>), 117.0 (C<sup>8</sup>,  ${}^{1}J_{C,F} = 276.8$ ,  ${}^{2}J_{C,F} = 34.4$  Hz), 116.8 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 276.2$ ,  ${}^{2}J_{C,F} = 34.5$  Hz), 106.7 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 267.5$ ,  ${}^{2}J_{C,F} = 34.4$  Hz), 106.3 (C<sup>10</sup>,  ${}^{1}J_{C,F} = 267.7, {}^{2}J_{C,F} = 34.2$  Hz), 108.9 (C<sup>6</sup>,  ${}^{1}J_{C,F} =$ 258.8,  ${}^{2}J_{C,F} = 31.6$  Hz), 108.5 (C<sup>9</sup>,  ${}^{1}J_{C,F} = 258.5$ ,  ${}^{2}J_{C,F} = 31.7$  Hz).  ${}^{19}F$  NMR spectrum,  $\delta_{F}$ , ppm: 82.5 t (3F, 11-F, J = 9 Hz), 83.8 t (3F, 8-F, J = 9 Hz),53.0 q (2F, 7-F, J = 9 Hz), 50.0 q (2F, 10-F, J =9 Hz), 38.2 s (2F, 6-F), 36.4 s (2F, 9-F). Mass spectrum, m/z ( $I_{\rm rel}$ , %): 481 (73)  $M^+$ , 462 (20)  $[M-F]^+$ , 362 (51)  $[M-C_2F_5]^+$ , 262 (2)  $[M-2C_2F_5]^+$ , 169 (2)  $[C_{3}F_{7}]^{+}$ , 119 (5)  $[C_{2}F_{5}]^{+}$ , 91 (100)  $[C_{6}H_{5}N]^{+}$ , 77 (18)  $[C_6H_5]^+$ , 69 (6)  $[CF_3]^+$ . Found: *M* 481.0245.  $C_{14}H_5F_{14}N_3$ . Calculated: *M* 481.0259.

**3,5-Bis(heptafluoropropyl)-1-(2-nitrophenyl)-1,2,4-triazole (III)** was synthesized from 5.2 g (34 mmol) of 2-nitrophenylhydrazine and 14.7 g (34 mmol) of compound **I**. Yield 14.8 g (83%); yellow liquid, bp 150–151°C (0.5 mm). IR spectrum, v, cm<sup>-1</sup>: 1610, 1590, 1545 (C=C<sub>arom</sub>); 1520 (C=C-C=N); 1540 (NO<sub>2</sub>, asym.); 1350 (NO<sub>2</sub>, sym.); 1320 (C-N); 1220–1150 (C-F). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: 8.31 (17-H), 7.88 (15-H, 16-H), 7.71 (14-H). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{C}$ , ppm: 154.5 (C<sup>3</sup>,  ${}^{2}J_{C,F} = 29.6$  Hz), 146.6 (C<sup>5</sup>,  ${}^{2}J_{C,F} =$ 30.5 Hz), 159.9 (C<sup>12</sup>), 144.2 (C<sup>13</sup>), 134.7 (C<sup>14</sup>), 133.1 (C<sup>15</sup>), 129.6 (C<sup>16</sup>), 126.1 (C<sup>17</sup>), 117.8 (C<sup>8</sup>,  ${}^{1}J_{C,F} =$ 286.9,  ${}^{2}J_{C,F} = 30$  Hz), 117.5 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 287.2$ ,  ${}^{2}J_{C,F} = 29.6$  Hz), 110.5 (C<sup>6</sup>,  ${}^{1}J_{C,F} = 256.8$ ,  ${}^{2}J_{C,F} =$ 28.7 Hz), 110.4 (C<sup>9</sup>,  ${}^{1}J_{C_1F} = 259.0$ ,  ${}^{2}J_{C,F} = 31$  Hz), 108.6 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 266.9$ ,  ${}^{2}J_{C,F} = 27.8$  Hz), 108.4 (C<sup>10</sup>,  ${}^{1}J_{C,F} = 268.7$ ,  ${}^{2}J_{C,F} = 38.4$  Hz).  ${}^{19}F$  NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 82.9 t (3F, 8-F, J = 9 Hz), 82.5 t (3F, 11-F, J = 9 Hz), 53.8 q (2F, 7-F, J = 9 Hz),49.5 q (2F, 10-F, J 9 Hz), 37.6 s (2F, 6-F), 36.1 s (2F, 9-F). Mass spectrum, m/z ( $I_{rel}$ , %): 526 (6)  $M^+$ , 507 (49)  $[M-F]^+$ , 480 (2)  $[M-NO_2]^+$ , 457 (1)  $[M-CF_3]^+$ , 407 (27)  $[M-C_2F_5]^+$ , 195 (1)  $[C_3F_7CN]^+$ , 169 (54)  $[C_{3}F_{7}]^{+}$ , 150 (4)  $[C_{6}H_{4}NO_{2}N_{2}]^{+}$ , 119 (9)  $[C_{2}F_{5}]^{+}$ , 118 (1)  $[C_6H_4NO_2]^+$ , 106 (87)  $[C_6H_4NO]^+$ , 100 (5)  $[CF_2=CF_2]^+$ , 90 (32)  $[C_6H_4N]^+$ , 78 (100)  $[C_6H_6]^+$ , 76 (17)  $[C_6H_4]^+$ , 69 (50)  $[CF_3]^+$ , 46 (1)  $[NO_2]^+$ . Found: M 526.0124.  $C_{14}H_4F_{14}N_4O_2$ . Calculated: *M* 526.0110.

**1-(2,4-Dinitrophenyl)-3,5-bis(heptafluoropropyl)-1,2,4-triazole (IV)** was synthesized from 5.03 g (25 mmol) of 2,4-dinitrophenylhydrazine and 10.8 g (25 mmol) of compound **I**. Yield 7.5 g (53%); brown liquid, bp 104–106°C (0.3 mm). UV spectrum (C<sub>2</sub>H<sub>5</sub>OH),  $\lambda_{max}$ , nm (ε, 1 mol<sup>-1</sup> cm<sup>-1</sup>): 228 (11800), 334 (850). IR spectrum, v, cm<sup>-1</sup>: 1590, 1545 (C=C<sub>arom</sub>); 1610, 1520 (C=C-C=N); 1540 (NO<sub>2</sub>, asym.); 1350 (NO<sub>2</sub>, sym.); 1320 (C-N); 1220–1150 (C-F). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 9.15 d (14-H, J = 2.5 Hz), 8.78 d.d (17-H, J = 8, 2.5 Hz), 8.02 d (16-H, J = 8 Hz). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{C}$ , ppm: 155.1 (C<sup>3</sup>, <sup>2</sup>J<sub>C,F</sub> = 31 Hz), 146.7 (C<sup>5</sup>, <sup>2</sup>J<sub>C,F</sub> = 30 Hz), 145.0 (C<sup>12</sup>), 134 (C<sup>13</sup>, C<sup>15</sup>), 129.1 (C<sup>16</sup>), 129.0 (C<sup>14</sup>), 122 (C<sup>17</sup>), 117.3 (C<sup>8</sup>, <sup>1</sup>J<sub>C,F</sub> = 288.1, <sup>2</sup>J<sub>C,F</sub> = 32.8 Hz), 117.6 (C<sup>11</sup>, <sup>1</sup>J<sub>C,F</sub> = 287.3, <sup>2</sup>J<sub>C,F</sub> = 33.1 Hz), 108.2 (C<sup>7</sup>, <sup>1</sup>J<sub>C,F</sub> = 263.4, <sup>2</sup>J<sub>C,F</sub> = 34.6 Hz), 110.1 (C<sup>6</sup>, <sup>1</sup>J<sub>C,F</sub> = 262.4, <sup>2</sup>J<sub>C,F</sub> = 31.2 Hz), 110.2 (C<sup>7</sup>, <sup>1</sup>J<sub>C,F</sub> = 257.4, <sup>2</sup>J<sub>C,F</sub> = 31.2 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 82.9 t (3F, 8-F, J = 9 Hz), 82.5 t (3F, 11-F, J = 9 Hz), 54.0 q (2F, 7-F, J = 9 Hz), 49.2 q (2F, 10-F, J = 9 Hz), 37.7 s (2F, 6-F), 36.0 s (2F, 9-F). Mass spectrum, m/z ( $I_{\rm rel}$ , %): 571 (3)  $M^+$ , 552 (42)  $[M-F]^+$ , 452 (37)  $[M-C_2F_5]^+$ , 404 (2)  $[M-C_6H_3(NO_2)_2]^+$ , 376 (36)  $[M-C_3F_7CN]^+$ , 348 (17)  $[C_3F_7CN, N_2]^+$ , 195 (1)  $[C_3F_7CN]^+$ , 181 (4)  $[C_6H_3(NO_2)_2N]^+$ , 169 (100)  $[C_3F_7]^+$ , 145 (2)  $[C_2F_5CN]^+$ , 119 (11)  $[C_2F_5]^+$ , 100 (9)  $[CF_2=CF_2]^+$ , 69 (49)  $[CF_3]^+$ . Found, %: C 29.42; H 0.53; F 46.58; N 12.26. M 570.9953.  $C_{14}H_3F_{14}N_5O_4$ . Calculated, %: C 29.34; H 0.45; F 46.68; N 12.47. M 570.9961.

**Perfluoro**(3,5-dipropyl-1-phenyl-1,2,4-triazole) (V) was synthesized from 4.95 g (25 mmol) of pentafluorophenylhydrazine and 10.8 g (25 mmol) of compound I. Yield 12.8 g (90%); yellow liquid, bp 71-72°C (3 mm). UV spectrum (C<sub>2</sub>H<sub>5</sub>OH),  $\lambda_{max}$ , nm ( $\epsilon$ , 1 mol<sup>-1</sup> cm): 265 (900). IR spectrum, v, cm<sup>-1</sup>: 1527 (C=C), 1520 (C=N), 1338 (C-N), 1236-1126 (C-F). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 155.9 (C<sup>3</sup>,  ${}^{2}J_{C,F} = 30.0$  Hz), 147.9 (C<sup>5</sup>,  ${}^{2}J_{C,F} = 30.9$  Hz), 144.4 (C<sup>15</sup>,  ${}^{1}J_{C,F} = 253.9$  Hz), 144 (C<sup>13</sup>,  ${}^{1}J_{C,F} = 258.7$  Hz), 138.3 (C<sup>14</sup>,  ${}^{1}J_{C,F} = 258.2$  Hz), 117.9 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 261.7, {}^{2}J_{C,F} = 33.2$  Hz), 117.4 (C<sup>8</sup>,  ${}^{1}J_{C,F} =$ 253.5,  ${}^{2}J_{C,F} = 33.5 \text{ Hz}$ , 111.5 (C<sup>12</sup>,  ${}^{2}J_{C,F} = 35.8 \text{ Hz}$ ), 110.2 ( $C^{9}$ ,  ${}^{1}J_{C,F} = 258.5$ ,  ${}^{2}J_{C,F} = 32.1$  Hz), 109.5 ( $C^{6}$ ,  ${}^{1}J_{C,F} = 257.9, {}^{2}J_{C,F} = 34.1$  Hz), 108.9 (C<sup>10</sup>,  ${}^{1}J_{C,F} =$ 267.4,  ${}^{2}J_{C,F} = 34.1$  Hz), 108.3 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 268$ ,  $^{2}J_{C E} = 33.9$  Hz).  $^{19}F$  NMR spectrum (CDCl<sub>3</sub>),  $\delta_{E}$ , ppm: 83.5 s (3F, 11-F), 83.1 s (3F, 8-F), 52.6 m (2F, 10-F), 49.7 m (2F, 7-F), 38 m (2F, 9-F), 36.8 m (2F, 6-F), 19.3 m (2F, 13-F), 16.6 m (1F, 15-F), 3.4 m (2F, 14-F). Mass spectrum, m/z ( $I_{rel}$ , %): 571  $M^+$ , 552  $[M-F]^+$ , 452  $[M-C_2F_5]^+$ , 376  $[M-C_3F_7CN]^+$ , 357  $[M-C_2F_5CF=C=N]^+$ , 257, 224, 169  $[C_3F_7]^+$ , 167  $[C_6F_5]^+$ , 119  $[C_2F_5]^+$ , 69  $[CF_3]^+$ . Found: *M* 570.97830.  $C_{14}F_{19}N_3$ . Calculated: *M* 570.97886.

**3,5-Bis(heptafluoropropyl)-1-(2,3,5,6-tetrafluorophenyl)-1,2,4-triazole (VI)** was synthesized from 1.8 g (10 mmol) of 2,3,5,6-tetrafluorophenylhydrazine, 4.33 g (10 mmol) of compound **I**, and 2.02 g (20 mmol) of triethylamine. Yield 4 g (72%); yellow liquid, bp 134–135°C (3 mm). UV spectrum (EtOH),  $\lambda_{\text{max}}$ , nm ( $\varepsilon$ , 1 mol<sup>-1</sup> cm<sup>-1</sup>): 271 (2700). IR spectrum, v, cm<sup>-1</sup>: 3086 (C–N), 1650 (C=C), 1631 (C=N), 1531 and 1508 (C=C<sub>arom</sub>), 1345 (C–N), 1236–1187 (C–F). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.47 (H<sub>arom</sub>). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\text{C}}$ , ppm: 156.1 (C<sup>3</sup>, <sup>2</sup>J<sub>C,F</sub> = 30.1 Hz), 148.1 (C<sup>15</sup>, <sup>2</sup>J<sub>C,F</sub> = 26.4 Hz), 146.3 (C<sup>13</sup>, <sup>1</sup>J<sub>C,F</sub> = 252.7, <sup>2</sup>J<sub>C,F</sub> = 15.7 Hz), 143.5 (C<sup>14</sup>,

 ${}^{1}J_{C,F} = 259.2, {}^{2}J_{C,F} = 15.9 \text{ Hz}$ , 118.3 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 285.7, {}^{2}J_{C,F} = 33.2 \text{ Hz}$ ), 117.9 (C<sup>8</sup>,  ${}^{1}J_{C,F} = 283.5$ ,  ${}^{2}J_{C,F} = 33.5 \text{ Hz}$ ), 110.3 (C<sup>12</sup>,  ${}^{2}J_{C,F} = 14.2 \text{ Hz}$ ), 110.2 (C<sup>9</sup>,  ${}^{1}J_{C,F} = 258.5, {}^{2}J_{C,F} = 32.1 \text{ Hz}$ ), 109.5 (C<sup>6</sup>,  ${}^{1}J_{C,F} = 257.9, {}^{2}J_{C,F} = 34.1 \text{ Hz}$ ), 108.9 (C<sup>10</sup>,  ${}^{1}J_{C,F} = 267.4, {}^{2}J_{C,F} = 34.1 \text{ Hz}$ ), 108.3 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 268, {}^{2}J_{C,F} = 33.9 \text{ Hz}$ ).  ${}^{19}\text{F}$  NMR spectrum (CDCl<sub>3</sub>),  $\delta_{F}$ , ppm: 83.4 s (3F, 11-F), 83.0 s (3F, 8-F), 52.3 m (2F, 10-F), 49.8 m (2F, 7-F), 37.8 m (2F, 9-F), 36.7 m (2F, 6-F), 27.1 m (2F, 13-F), 18.3 m (1F, 15-F). Mass spectrum, m/z ( $I_{rel}$ , %): 553 (50)  $M^+$ , 534 (32)  $[M-F]^+$ , 434 (78)  $[M-C_2F_5]^+$ , 239 (0.9)  $[M-C_3F_7\text{CN} - C_2F_5]^+$ , 215 (0.7)  $[\text{HC}_6F_4\text{N}_3\text{C}_2]^+$ , 177 (2)  $[\text{HC}_6F_4\text{N}_2]^+$ , 163 (100)  $[\text{HC}_6F_4\text{N}]^+$ , 149 (12)  $[\text{HC}_6F_4]^+$ , 119 (4)  $[C_2F_5]^+$ , 100 (3)  $[\text{CF}_2=\text{CF}_2]^+$ , 69 (22)  $[\text{CF}_3]^+$ . Found: M 552.9873.  $\text{C}_{14}\text{HF}_{18}\text{N}_3$ . Calculated: M 552.9883.

1-(4-Bromo-2,3,5,6-tetrafluorophenyl)-3,5-bis-(heptafluoropropyl)-1,2,4-triazole (VII) was synthesized from 2.59 g (10 mmol) of 4-bromo-2,3,5,6tetrafluorophenylhydrazine, 4.33 g (10 mmol) of compound I, and 2.02 g (20 mmol) of triethylamine. Yield 3.6 g (57%); yellow liquid, bp 90-91°C (2 mm). UV spectrum (EtOH),  $\lambda_{max}$ , nm ( $\epsilon$ , 1 mol<sup>-1</sup> cm<sup>-1</sup>): 250 (3500). IR spectrum, v, cm<sup>-1</sup>: 1711 (C=C), 1673 (C=N), 1527 and 1508 (C=C<sub>arom</sub>), 1404 and 1338 (C–N), 1229–1158 (C–F). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 161.4 (C<sup>3</sup>, <sup>2</sup>J<sub>C,F</sub> = 24.6 Hz), 159.1  $(C^5, {}^2J_{C,F} = 26.0 \text{ Hz}), 155.1 (C^{15}, {}^2J_{C,F} = 25.3 \text{ Hz}), 144.8 (C^{13}, {}^1J_{C,F} = 246.7, {}^2J_{C,F} = 15.7 \text{ Hz}), 142.8$  $(C^{14}, {}^{1}J_{C,F} = 257.2, {}^{2}J_{C,F} = 15.9 \text{ Hz}), 118.4 (C^{11},$  ${}^{1}J_{C,F} = 285.1, {}^{2}J_{C,F} = 33.8$  Hz), 117.9 (C<sup>8</sup>,  ${}^{1}J_{C,F} =$ 283.5,  ${}^{2}J_{C,F} = 33.5 \text{ Hz}$ , 110.3 (C<sup>12</sup>,  ${}^{2}J_{C,F} = 13.8 \text{ Hz}$ ), 110.2 ( $C^{9}$ ,  ${}^{1}J_{C,F} = 258.5$ ,  ${}^{2}J_{C,F} = 32.1$  Hz), 109.5 ( $C^{6}$ ,  ${}^{1}J_{C,F}$  257.5,  ${}^{2}J_{C,F}$  34.0 Hz), 108.9 (C<sup>10</sup>,  ${}^{1}J_{C,F}$  = 267.4,  ${}^{2}J_{C,F} = 34.1 \text{ Hz}$ , 108.3 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 268$ ,  ${}^{2}J_{C,F} =$ 33.9 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 82.8 s (3F, 11-F), 82.8 s (3F, 8-F), 52.2 m (2F, 10-F), 46.8 m (2F, 7-F), 37.8 m (2F, 9-F), 37.7 m (2F, 6-F), 18.8 m (2F, 13-F), 33.3 m (1F, 15-F). Mass spectrum, m/z ( $I_{\rm rel}$ , %): 631 (53)  $M^+$ , 612 (20)  $[M-F]^+$ , 512 (40)  $[M - C_2F_5]^+$ , 256 (4)  $[BrC_6F_4N_3]^+$ , 241 (44)  $[BrC_6F_4NH]^+$ , 229 (4)  $[BrC_6F_4]^+$ , 169 (12)  $[C_3F_7]^+$ , 148 (9)  $[C_6F_4]^+$ , 119 (10)  $[C_2F_5]^+$ , 100 (5)  $[CF_2=CF_2]^+$ , 69 (50)  $[CF_3]^+$ . Found: *M* 630.9396.  $C_{14}BrF_{18}N_3$ . Calculated: *M* 630.8988.

**3,5-Bis(heptafluoropropyl)-1-(2,3,5,6-tetrafluoro-4-trifluoromethylphenyl)-1,2,4-triazole (VIII)** was synthesized from 6.2 g (25 mmol) of 2,3,5,6-tetra-

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 37 No. 11 2001

fluoro-4-trifluoromethylphenylhydrazine and 10.8 g (25 mmol) of compound I. Yield 13 g (84%); yellow liquid, bp 80-81°C (3 mm). UV spectrum (EtOH),  $λ_{max}$ , nm (ε, 1 mol<sup>-1</sup> cm<sup>-1</sup>): 282 (2300). IR spectrum, v, cm<sup>-1</sup>: 1658 (C=C), 1514 (C=N), 1510 (C=C<sub>arom</sub>), 1340 (C-N), 1220–1124 (C-F). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 156.9 (C<sup>3</sup>,  ${}^{2}J_{\rm C,F}$  = 30.3 Hz), 148.3 (C<sup>5</sup>,  ${}^{2}J_{\rm C,F}$  = 31.2 Hz), 145 (C<sup>14</sup>,  ${}^{1}J_{\rm C,F}$  = 269.4 Hz), 144.1 (C<sup>13</sup>,  ${}^{1}J_{C,F} = 261.9$  Hz), 136.2 (C<sup>15</sup>,  ${}^{2}J_{C,F} = 32$  Hz), 120.7 (C<sup>16</sup>,  ${}^{1}J_{C,F} = 275.5$  Hz), 118.2 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 286.8, {}^{2}J_{C,F} = 33.4$  Hz), 117.8 (C<sup>8</sup>,  ${}^{1}J_{C,F} =$ 286.2,  ${}^{2}J_{C,F} = 33.5$  Hz), 110.7 (C<sup>9</sup>,  ${}^{1}J_{C,F} = 259.5$ ,  ${}^{2}J_{C,F} = 40.2$  Hz), 110.6 (C<sup>6</sup>,  ${}^{1}J_{C,F} = 257.7$ ,  ${}^{2}J_{C,F} = 39.2$  Hz), 108.9 (C<sup>10</sup>,  ${}^{1}J_{C,F} = 267.0$ ,  ${}^{2}J_{C,F} = 38.6$  Hz), 108.8 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 268.5$ ,  ${}^{2}J_{C,F} = 39.7$  Hz).  ${}^{19}F$  NMR spectrum (CDCl<sub>3</sub>), δ<sub>F</sub>, ppm: 106.8 s (3F, 16-F), 83.5 s (3F, 11-F), 83.1 s (3F, 8-F), 52.5 m (2F, 10-F), 49.7 m (2F, 7-F), 38 m (2F, 9-F), 36.8 m (2F, 6-F), 26.2 m (2F, 13-F), 21.3 m (2F, 14-F). Mass spectrum, m/z ( $I_{rel}$ , %): 621  $M^+$ , 602  $[M-F]^+$ , 502  $[M-C_2F_5]^+$ , 426  $[M-C_3F_7CN]^+$ , 407  $[M-C_2F_5CF=C=N]^+$ , 306, 231, 216, 169  $[C_3F_7]^+$ , 69  $[CF_3]^+$ . Found: *M* 620.97550. C<sub>15</sub>F<sub>21</sub>N<sub>3</sub>. Calculated: *M* 620.97567.

**4,4'-Bis[3,5-bis(heptafluoropropyl)-1,2,4-triazol-1-yl]octafluorobiphenyl (IX)** was synthesized from 4.1 g (11.5 mmol) of 4,4'-dihydrazinooctafluorobiphenyl and 10 g (23 mmol) of compound **I**. Yield 5.4 g (42%); yellow liquid, bp 163–164°C (0.2 mm). UV spectrum (EtOH),  $\lambda_{max}$ , nm ( $\varepsilon$ , 1 mol<sup>-1</sup> cm<sup>-1</sup>): 240 (8200), 280 (1750). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{C}$ , ppm: 156.9 (C<sup>3</sup>, <sup>2</sup>*J*<sub>C,F</sub> = 30.2 Hz), 148.5 (C<sup>5</sup>, <sup>2</sup>*J*<sub>C,F</sub> = 30.9 Hz), 144.9 (C<sup>13</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 251.9, <sup>2</sup>*J*<sub>C,F</sub> = 12.5 Hz), 144.1 (C<sup>14</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 261.7, <sup>2</sup>*J*<sub>C,F</sub> = 12.5 Hz), 118.8 (C<sup>12</sup>, <sup>2</sup>*J*<sub>C,F</sub> = 24.4 Hz), 118.5 (C<sup>8</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 286.8, <sup>2</sup>*J*<sub>C,F</sub> = 33.2 Hz), 118.1 (C<sup>11</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 287.1, <sup>2</sup>*J*<sub>C,F</sub> = 33.1 Hz), 110.7 (C<sup>15</sup>, <sup>2</sup>*J*<sub>C,F</sub> = 24.6 Hz), 110.5 (C<sup>6</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 256.8, <sup>2</sup>*J*<sub>C,F</sub> = 31.3 Hz), 110.4 (C<sup>9</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 259.0, <sup>2</sup>*J*<sub>C,F</sub> = 31 Hz), 108.9 (C<sup>7</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 268.9, <sup>2</sup>*J*<sub>C,F</sub> = 32.9 Hz), 108.4 (C<sup>10</sup>, <sup>1</sup>*J*<sub>C,F</sub> = 268.7, <sup>2</sup>*J*<sub>C,F</sub> = 30.4 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{F}$ , ppm: 83.5 t (8-F, *J* = 9 Hz), 83.1 t (11-F, *J* = 9 Hz), 52.5 q (7-F, *J* = 9 Hz), 49.7 q (10-F, *J* = 9 Hz), 37.9 s (6-F), 36.6 s (9-F), 29.1 m (14-F), 19.9 m (13-F). Found, %: C 30.65; F 61.85; N 7.30. C<sub>28</sub>F<sub>36</sub>N<sub>6</sub>. Calculated, %: C 30.43; F 61.96; N 7.61.

**5-Fluoro-3-(pentafluoroethyl)-1-(2,3,5,6-tetrafluorophenyl)-4-trifluoromethyl-1***H***-pyrazole (XX) was synthesized from 3 g (10 mmol) of compound XIX, 1.8 g (10 mmol) of 2,3,5,6-tetrafluorophenylhydrazine, and 3.03 g (30 mmol) of triethylamine.** 

Yield 3.5 g (83%); slightly yellowish liquid, bp 69– 70°C. IR spectrum, v, cm<sup>-1</sup>: 3084 (C-N), 1620 (C=C), 1527 (C=C<sub>arom</sub>), 1471 and 1887 (C-N), 1225–1160 (C-F). <sup>1</sup>H NMR spectrum, δ, ppm: 7.44. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 141.0 (C<sup>3</sup>, <sup>2</sup> $J_{C,F} = 31.5$  Hz), 146.7 (C<sup>11</sup>, <sup>1</sup> $J_{C,F} = 257.3$ , <sup>2</sup> $J_{C,F} = 13.5$  Hz), 143.4 (C<sup>10</sup>, <sup>1</sup> $J_{C,F} = 258.8$ , <sup>2</sup> $J_{C,F} = 15.5$  Hz), 120 (C<sup>6</sup>, <sup>1</sup> $J_{C,F} = 267.8$  Hz), 118.7 (C<sup>8</sup>, <sup>1</sup> $J_{C,F} = 286.3$ ,  ${}^{2}J_{C,F} = 41.4$  Hz), 110.3 (C<sup>9</sup>,  ${}^{2}J_{C,F} = 13.4$  Hz), 109.4  $(C^{7}, {}^{1}J_{C,F} = 253.8, {}^{2}J_{C,F} = 40.3 \text{ Hz}), 109.4 (C^{12}, {}^{2}J_{C,F} = 15.5 \text{ Hz}), 96.0 (C^{4}, {}^{2}J_{C,F} = 34.3, {}^{2}J_{C,F} =$ 8.5 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 107.1 d.t (6-F, J = 22, 7 Hz), 80.2 s (8-F), 51.8 q (7-F, J = 7 Hz), 43.4 q (5-F, J = 22 Hz), 27.1 m(11-F), 17.4 m (10-F). Mass spectrum, m/z ( $I_{rel}$ , %): 420 (59)  $M^+$ , 401 (29)  $[M-F]^+$ , 351 (100)  $[M-CF_3]^+$ , 301 (8)  $[M-C_2F_5]^+$ , 282 (10)  $[M-F-C_2F_5]^+$ , 263 (5)  $[M-2F-C_{2}F_{5}]^{+}$ , 256 (8)  $[M-HC_{6}F_{4}N]^{+}$ , 232 (0.7)  $[M-CF_3-C_2F_5]^+$ , 207 (0.4)  $[M-HC_6F_4NCF_2]^+$ , 176 (7)  $[C_6F_4N_2]^+$ , 149 (12)  $[HC_6F_4]^+$ , 119 (3)  $[C_2F_5]^+$ , 100 (2)  $[CF_2=CF_2]^+$ , 69 (28)  $[CF_3]^+$ . Found: *M* 419.9967. C<sub>12</sub>HF<sub>13</sub>N<sub>2</sub>. Calculated: *M* 419.9932.

1-(4-Bromo-2,3,5,6-tetrafluorophenyl)-5-fluoro-3-pentafluoroethyl-4-trifluoromethyl-1*H*-pyrazole (XXI) was synthesized from 3 g (10 mmol) of compound XIX, 2.59 g (10 mmol) of 4-bromo-2,3,5,6tetrafluorophenylhydrazine, and 3.03 g (30 mmol) of triethylamine. Liquid substance, bp 84–85°C (2 mm). IR spectrum, v, cm<sup>-1</sup>: 1623 (C=C), 1539 (C=N), 1503 (C=C<sub>arom</sub>), 1357 and 1328 (C-N), 1225–1162 (C-F). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{C}$ , ppm: 152.3  $(C^5, {}^1J_{C,F} = 298.8 \text{ Hz}), 141.3 (C^3, {}^2J_{C,F} = 39.6 \text{ Hz}),$ 145.7 (C<sup>11</sup>,  ${}^{1}J_{C,F} = 255.3$ ,  ${}^{2}J_{C,F} = 13.5$  Hz), 144.4  $(C^{10}, {}^{1}J_{C,F} = 263.3, {}^{2}J_{C,F} = 15.5 \text{ Hz}), 119.6 (C^{6},$  ${}^{1}J_{C,F} = 272.8 \text{ Hz}$ , 118.2 (C<sup>8</sup>,  ${}^{1}J_{C,F} = 286.1$ ,  ${}^{2}J_{C,F} = 36.4 \text{ Hz}$ ), 115.8 (C<sup>12</sup>,  ${}^{2}J_{C,F} = 15.5 \text{ Hz}$ ), 109.7 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 267.9, {}^{2}J_{C,F} = 40.3 \text{ Hz}), 104.5 (C^{9}, {}^{2}J_{C,F} = 22.5 \text{ Hz}), 95.4 (C^{4}, {}^{2}J_{C,F} = 34.3, {}^{2}J_{C,F} = 8.5 \text{ Hz}).$ <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 106.7 d.t (6-F, J = 22, 7 Hz), 80.4 s (8-F), 51.4 q (7-F, J = 7 Hz), 44.1 q (5-F, *J* = 22 Hz), 32.9 m (11-F), 18.7 m (10-F). Mass spectrum, m/z ( $I_{rel}$ , %): 498 (100)  $M^+$ , 479 (32)  $[M-F]^+$ , 429 (93)  $[M-CF_3]^+$ , 256 (2)  $[BrC_6F_4N_2]^+$ , 228 (1)  $[BrC_6F_4]^+$ , 119 (14)  $[C_2F_5]^+$ , 100 (3)  $[CF_2=CF_2]^+$ , 69 (66)  $[CF_3]^+$ . Found: *M* 497.9028.  $C_{12}BrF_{13}N_2$ . Calculated: *M* 497.9038.

**Reactions of compounds I and XIX with hydrazine hydrate** (*general procedure*). Hydrazine hydrate, 2 equiv, was added at 0°C to a solution of compound I or XIX in 35 ml of acetic acid. The mixture was stirred for 1 h at that temperature, heated for 2 h at  $45^{\circ}$ C, cooled, poured into 100 ml of water, and treated with methylene chloride (3 × 70 ml). The extract was washed with water (2 × 50 ml), dried over MgSO<sub>4</sub>, and evaporated, and the residue was distilled or subjected to column chromatography on silica gel using hexane–methylene chloride as eluent.

**3,5-Bis(heptafluoropropyl)-4***H***-1,2,4-triazole** (XV) was synthesized from 17 g (39 mmol) of perfluoro(5-aza-4-nonene) (I) and 3.9 g (78 mmol). of hydrazine hydrate. Yield 6.7 g (42%), mp 96-98°C. UV spectrum (EtOH),  $\lambda_{max}$ , nm ( $\epsilon$ , 1 mol<sup>-1</sup> cm<sup>-1</sup>): 334 (4200). IR spectrum, v, cm<sup>-1</sup>: 1620 (C=N), 1450 (C=N), 1350 (C-N), 1120–1220 (C-F). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 8.17 (NH). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 158.7 (C<sup>3</sup>, C<sup>5</sup>, <sup>2</sup>J<sub>C,F</sub> = 25.7 Hz), 117.3 (C<sup>8</sup>, C<sup>11</sup>, <sup>1</sup>J<sub>C,F</sub> = 287.3, <sup>2</sup>J<sub>C,F</sub> = 33.5 Hz), 110.2 (C<sup>6</sup>, C<sup>9</sup>, <sup>1</sup>J<sub>C,F</sub> = 264.4, <sup>2</sup>J<sub>C,F</sub> = 31.4 Hz), 108.0 (C<sup>7</sup>,  $C^{10}$ ,  ${}^{1}J_{C,F} = 255.5$ ,  ${}^{2}J_{C,F} = 31.8$  Hz).  ${}^{19}F$  NMR spectrum (CDCl<sub>3</sub>),  $\delta_{F}$ , ppm: 82.97 (6F, 8-F, 11-F), 50.69 (4F, 7-F, 10-F), 36.8 (4F, 6-F, 9-F). Mass spectrum, m/z ( $I_{\rm rel}$ , %): 405 (1)  $M^+$ , 386 (26)  $[M-F]^+$ , 336 (1)  $[M-CF_3]^+$ , 286 (100)  $[M-C_2F_5]^+$ , 236 (1)  $[M-C_2F_5]^+$  $(C_3F_7)^+$ , 196 (1)  $[C_3F_7C=NH]^+$ , 169 (5)  $[C_3F_7]^+$ , 119 (5)  $[C_2F_5]^+$ , 76 (1)  $[CF_2=C=N]^+$ , 69 (33)  $[CF_3]^+$ . Found: *M* 404.9859. C<sub>8</sub>HF<sub>14</sub>N<sub>3</sub>. Calculated: M 404.9834.

3,3,3-Trifluoro-1-pentafluoroethyl-2-trifluoromethylpropylidenehydrazine XXIIIa/XXIIIb. By the above procedure, from 9 g (30 mmol) of compound XIX and 3 g (60 mmol) of hydrazine hydrate we obtained a mixture of isomeric hydrazones XXIIIa and XXIIIb at a ratio of 2.1:1 (GLC) as a yellow liquid, bp 128–129°C. Yield 5 g (53%). IR spectrum, v, cm<sup>-1</sup>: 3400, 3500 (NH<sub>2</sub>); 2990 (C-H); 1710 (C=N); 1600 (C=N); 1380, 1350 (C-N); 1280–1150 (C-F). Mass spectrum, m/z ( $I_{rel}$ , %): 312 (68)  $M^+$ , 293 (17)  $[M-F]^+$ , 273 (6)  $[M-F-HF]^+$ , 243 (5)  $[M - CF_3]^+$ , 193 (100)  $[M - C_2F_5]^+$ , 161 (23)  $[C_2F_5C=NNH_2]^+$ , 119 (11)  $[C_2F_5]^+$ , 100 (1)  $[CF_2=CF_2]^+$ , 69 (28)  $[CF_3]^+$ , 42 (1)  $[H_2NN=C]^+$ . Found, %: C 24.77; H 1.04; F 67.18; N 8.50. *M* 312.0127.  $C_6H_3F_{11}N_2$ . Calculated, %: C 23.08; H 0.96; F 66.99; N 8.97. M 312.0121.

Isomer **XXIIIa**. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.67 (CH), 6.30 (NH<sub>2</sub>). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 165.8 (C<sup>3</sup>, <sup>2</sup>J<sub>C,F</sub> = 27.0 Hz), 113.7 (C<sup>1</sup>, C<sup>6</sup>, <sup>1</sup>J<sub>C,F</sub> = 282.1 Hz), 110.4 (C<sup>5</sup>, <sup>1</sup>J<sub>C,F</sub> = 288.1, <sup>2</sup>J<sub>C,F</sub> = 36.4 Hz), 101.8 (C<sup>4</sup>, <sup>1</sup>J<sub>C,F</sub> = 260.8, <sup>2</sup>J<sub>C,F</sub> = 41.8 Hz), 38.0 (C<sup>2</sup>, <sup>2</sup>J<sub>C,F</sub> = 32.5 Hz). <sup>19</sup>F NMR spectrum

 $(CDCl_3)$ ,  $\delta_F$ , ppm: 102.2 (6F, 1-F, 6-F), 82.0 (3F, 5-F), 52.3 (2F, 4-F).

Isomer **XXIIIb.** <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.67 (CH), 6.10 (NH<sub>2</sub>). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 165.4 (C<sup>3</sup>, <sup>2</sup>J<sub>C,F</sub> = 26.8 Hz), 113.6 (C<sup>1</sup>, C<sup>6</sup>, <sup>1</sup>J<sub>C,F</sub> = 282.1 Hz), 110.1 (C<sup>5</sup>, <sup>1</sup>J<sub>C,F</sub> = 282.9, <sup>2</sup>J<sub>C,F</sub> = 36.6 Hz), 103.6 (C<sup>4</sup>, <sup>1</sup>J<sub>C,F</sub> = 260.5, <sup>2</sup>J<sub>C,F</sub> = 34.7 Hz), 35.6 (C<sup>2</sup>, <sup>2</sup>J<sub>C,F</sub> = 31.2 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 98.3 (6F, 1-F), 79.8 (3F, 5-F), 46.7 (2F, 4-F).

Bis(trifluoromethyl)bis[2,2,2-trifluoro-1-(trifluoromethyl)ethyl]cyclobutadienes XXIVa and **XXIVb.** A mixture of 6.5 g (21 mmol) of compounds **XXIIIa/XXIIIb** and 4.2 g (42 mmol) of triethylamine in 30 ml of acetonitrile was stirred for 3 h at 80°C. It was then poured into 100 ml of water and treated with methylene chloride  $(3 \times 70 \text{ ml})$ . The extract was washed with water  $(2 \times 50 \text{ ml})$  and dried over Na<sub>2</sub>SO<sub>4</sub>. Removal of the solvent, followed by distillation, gave 3.2 g (63%) of a mixture of compounds **XXIVa** and **XXIVb** (ratio 2.3:1, according to the GLC data) as a volatile mobile liquid, bp 80-82°C. Mass spectrum, m/z ( $I_{\rm rel}$ , %): 488 (17)  $M^+$ , 469 (28)  $[M-F]^+$ , 419 (8)  $[M-CF_3]^+$ , 400 (10)  $[M-CF_3-F]^+$ , 337 (2) [M- $CH(CF_3)_2]^+$ , 268 (1)  $[M-CF_3-CH(CF_3)_2]^+$ , 244 (1)  $[(CF_3)_2 CHCCCF_3]^+$ , 151 (1)  $[(CF_3)_2 CH]^+$ , 119 (7)  $[C_{2}F_{5}]^{+}$ , 100 (1)  $[CF_{2}=CF_{2}]^{+}$ , 82 (1)  $[CF_{3}CH]^{+}$ , 69 (83)  $[CF_3]^+$ , 41 (100), 40 (49), 39 (16). Found: *M* 487.9846. C<sub>12</sub>H<sub>2</sub>F<sub>18</sub>. Calculated: *M* 487.9869.

Isomer **XXIVa**. <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 129.5 (C<sup>2</sup>, C<sup>4</sup>), 120.3 (C<sup>7</sup>, C<sup>8</sup>, C<sup>11</sup>, C<sup>12</sup>, <sup>1</sup> $J_{\rm C,F}$  = 282.6 Hz), 119.8 (C<sup>5</sup>, C<sup>9</sup>, <sup>1</sup> $J_{\rm C,F}$  = 269.8 Hz), 113.2 (C<sup>1</sup>, C<sup>3</sup>, <sup>2</sup> $J_{\rm C,F}$  = 37.8 Hz), 46.6 (C<sup>6</sup>, C<sup>10</sup>, <sup>2</sup> $J_{\rm C,F}$  = 22.5 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 102.9 (6F, 5-F, 9-F), 96.8 (12F, 7-F, 8-F, 11-F, 12-F).

Isomer **XXIVb.** <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm C}$ , ppm: 120.4 (C<sup>8</sup>, C<sup>9</sup>, C<sup>11</sup>, C<sup>12</sup>, <sup>1</sup>J<sub>C,F</sub> = 283.2 Hz), 119.9 (C<sup>5</sup>, C<sup>6</sup>, <sup>1</sup>J<sub>C,F</sub> = 270.3 Hz), 129.6 (C<sup>3</sup>, C<sup>4</sup>), 114.2 (C<sup>1</sup>, C<sup>2</sup>, <sup>2</sup>J<sub>C,F</sub> = 38.0 Hz), 46.9 (C<sup>7</sup>, C<sup>10</sup>, <sup>2</sup>J<sub>C,F</sub> = 22.5 Hz). <sup>19</sup>F NMR spectrum (CDCl<sub>3</sub>),  $\delta_{\rm F}$ , ppm: 100.5 (12F, 8-F, 9-F, 11-F, 12-F), 103.5 (6F, 5-F, 6-F).

5-Fluoro-3-pentafluoroethyl-4-trifluoromethyl-1*H*-pyrazole (XXII). A solution of 15 g (50 mmol) of compound XIX and 10.1 g (100 mmol) of triethylamine in 40 ml of acetonitrile was stirred for 2 h at 40°C. It was then cooled to 0°C, and 5.1 g (50 mmol) of propionylhydrazine was added over a period of 15 min, maintaining the temperature at 0°C. The mixture was stirred for 1 h at 0°C and for 3 h at 40°C, 20 ml of DMF was added, and the mixture was heated

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 37 No. 11 2001

for 1 h at 40°C, cooled, poured into 100 ml of water, and extracted with methylene chloride  $(3 \times 70 \text{ ml})$ . The extract was washed with water  $(2 \times 50 \text{ ml})$ , dried over MgSO<sub>4</sub>, and evaporated, and the residue was distilled under reduced pressure. Yield 9.1 g (67%); vellow liquid, bp 60–61°C (0.6 mm). IR spectrum, v, cm<sup>-1</sup>: 3400 (NH); 1670 (C=N); 1500 (C=N); 1330, 1380 (C-N); 1140-1220 (C-F). <sup>1</sup>H NMR spectrum, δ, ppm: 8.24 (NH). <sup>13</sup>C NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>),  $\delta_{C}$ , ppm: 150.3 (C<sup>5</sup>,  ${}^{1}J_{C,F} = 259.0$  Hz), 122.6 (C<sup>3</sup>,  ${}^{2}J_{C,F} = 31.9$  Hz), 112.1 (C<sup>6</sup>,  ${}^{1}J_{C,F} = 266.6$ ,  ${}^{2}J_{C,F} = 5$  Hz), 109.6 (C<sup>8</sup>,  ${}^{1}J_{C,F} = 285.9$ ,  ${}^{2}J_{C,F} = 37.5$  Hz), 100.8 (C<sup>7</sup>,  ${}^{1}J_{C,F} = 254.9$ ,  ${}^{2}J_{C,F} = 40.7$  Hz), 86.9 (C<sup>4</sup>,  ${}^{2}J_{C,F} = 41.6$ ,  ${}^{2}J_{C,F} = 17.1$  Hz).  ${}^{19}F$  NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>),  $\delta_{F}$ , ppm: 108.1 d.t (3F, 6-F, J = 20, 10 Hz), 79.2 s (3F, 8-F), 51.10 q (2F, 7-F, J = 10 Hz), 36.1 q (1F, 5-F, J = 20 Hz). Mass spectrum, m/z ( $I_{rel}$ , %): 272 (47)  $M^+$ , 253 (44)  $[M-F]^+$ , 243 (1)  $[M-N_2-H]^+$ , 203 (100)  $[M-CF_3]^+$ , 183 (42)  $[M-CF_3-HF]^+$ , 155 (12)  $[M - CF_3 - HF - N_2]^+$ , 119 (12)  $[C_2F_5]^+$ , 69 (58) [CF<sub>3</sub>]<sup>+</sup>. Found, %: C 26.40; H 0.29; F 69.15; N 10.15. C<sub>6</sub>HF<sub>9</sub>N<sub>2</sub>. Calculated, %: C 26.47; H 0.37; F 68.87; N 10.29.

**3,5-Bis(heptafluoropropyl)-4***H***-1,2,4-triazole (<b>XV).** Propionylhydrazide, 3.06 g (30 mmol), was added with stirring at 0°C to a solution of 13 g (30 mmol) of compound **I** and 6.06 g (60 mmol) of triethylamine in 40 ml of acetonitrile. The mixture was stirred for 1 h at 0°C and for 3 h at 50°C, cooled, poured into 100 ml of water, and treated with CH<sub>2</sub>Cl<sub>2</sub> ( $3 \times 70$  ml). The extract was washed with water ( $2 \times 50$  ml) and dried over MgSO<sub>4</sub>. Distillation followed by column chromatography on silica gel (eluent CH<sub>2</sub>Cl<sub>2</sub>) gave 7.9 g (65%) of compound **XV**.

## REFERENCES

Furin, G.G., *Targets in Heterocyclic Systems (Chemistry and Properties)*, Attanasi, O.A. and Spinelli, D., Eds., Roma: Ital. Soc. Chem., 1998, vol. 2, pp. 355–441.

- Ritter, S.K. and Washington, C., Chem. Eng. News, 1995, vol. 73, pp. 39–44.
- Synthetic Fluorine Chemistry, Olah, G.A., Chambers, R.D., and Prakash, G.K.S., Eds., New York: Wiley, 1992; Tozer, M.J. and Herpin, T.F., *Tetrahedron*, 1996, vol. 52, no. 26, pp. 8619–8683.
- 4. Organofluorine Compounds in Medicinal Chemistry and Biomedical Applications, Filler, R., Kobayashi, Y., and Yagupolskii, L.M., Eds., Amsterdam: Elsevier, 1993.
- Bargamova, M.D., Karpavichyus, K.I., Belostotskii, A.M., Motsshikite, S.M., and Knunyants, I.L., USSR Inventor's Certificate no. 1456419, 1989; *Chem. Abstr.*, 1989, vol. 111, no. 97232 y.
- Bargamov, G.G., Mysov, E.I., and Bargamova, M.D., *Izv. Ross. Akad. Nauk, Ser. Khim.*, 1994, no. 11, pp. 2039–2040.
- 7. Bargamova, M.D. and German, L.S., *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1991, no. 6, pp. 1463–1464.
- Bargamova, M.D., Motsshikite, S.M., and Knunyants, I.L., USSR Inventor's Certificate no. 1456418, 1987; *Byull. Izobret.*, no. 5, p. 80.
- Popkova, V.Ya., Galakhov, M.V., and Knunyants, I.L., *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1989, no. 1, pp. 116–122.
- Ikeda, I., Kogame, Y., and Okahara, M., J. Org. Chem., 1985, vol. 50, no. 19, pp. 3640–3642.
- Miki, H. and Gogo, T., JPN Patent Appl. no. 63-162663, 1988; *Chem. Abstr.*, 1989, vol. 110, no. 7690 y.
- 12. Ikeda, I., Tsukamoto, T., and Okahara, M., *Chem. Lett.*, 1980, no. 3, p. 583.
- 13. Chi, K.-W., Kim, S.-J., Park, T.-H., Gatilov, Yu.V., Bagryanskaya, I.Yu., and Furin, G.G., *J. Fluorine Chem.*, 1999, vol. 98, no. 1, pp. 29–36.
- Chi, K.-W., Furin, G.G., Gatilov, Yu.V., and Bagryanskaya, I.Yu., *J. Fluorine Chem.*, 2000, vol. 103, no. 2, pp. 105–115.
- Petrov, V.A., Belen'kii, G.G., and German, L.S., *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1985, no. 8, pp. 1934– 1935.